

**Government of Maharashtra's  
Ismail Yusuf College of Arts, Science and Commerce, Mumbai  
400060**

NAAC reaccredited A grade

**Department of Chemistry**

**Sample Multiple Choice Questions**

1. Polymeric molecules \_\_\_\_\_ a definite crystalline structure.
  - a) have
  - b) do not have
  - c) completely having
  - d) partially having
  
2. As the crystallinity increases the brittleness of the polymer \_\_\_\_\_.
  - a) increases
  - b) decreases
  - c) moderate
  - d) remains constant
  
3. A polymeric molecule possess a molecular weight \_\_\_\_\_.
  - a) different
  - b) fixed
  - c) that can not be defined
  - d) may be determined
  
4. \_\_\_\_\_ is the property of recovering original shape after the removal of deforming strain.
  - a) rigidity modulus
  - b) young's modulus
  - c) elasticity
  - d) bulk modulus
  
5. The impact strength is measured as \_\_\_\_\_.
  - a) elasticity
  - b) strength
  - c) permeability
  - d) toughness
  
6. If the polymer is in the room temperature it is \_\_\_\_\_.

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- a) brittle
  - b) viscofluid state
  - c) amorphous
  - d) rubbery
7. The strength of the polymer increases with \_\_\_\_\_ in molecular weight.
- a) increase
  - b) decrease
  - c) no change
  - d) slightly decrease
8. Which of the following techniques is used to measure the number of conjugated double bonds and aromatic conjugation within the various polymer molecules?
- a) fourier transform infrared
  - b) differential scanning calorimetry
  - c) UV visible spectroscopy
  - d) thermo gravimetric analysis
9. Differential scanning calorimetry is useful for determining the \_\_\_\_\_.
- a) melting temperature, glass transition temperature, heat of fusion
  - b) volatilities of plasticizers and other additives
  - c) quantitative determination of additives in polymers
  - d) structural imperfections
10. Differential scanning calorimetry comes under the category of \_\_\_\_\_.
- a) spectral analysis
  - b) morphological analysis
  - c) thermal analysis
  - d) geological analysis
11. The study of fracture surfaces of polymeric material is done using \_\_\_\_\_.
- a) atomic force microscopy

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- b) x-ray diffraction
  - c) thermo gravimetric analysis
  - d) scanning electron microscopy
12. Elastic deformation in polymers is due to \_\_\_\_\_.
- a) slight adjust of molecular chains
  - b) slippage of molecular chains
  - c) straightening of molecular chains
  - d) severe of covalent bonds
13. Properties of polymer is affected by the \_\_\_\_\_.
- a) chain length
  - b) intermolecular forces
  - c) both a) and b)
  - d) none of these
14. Crystallisation of polymers is an undesirable property. Crystallisation of celluloid is prevented by adding \_\_\_\_\_
- a) glycerol
  - b) nitro cellulose
  - c) camphor
  - d) none of these
15. Visco-elastic behavior exhibited by plastics is a \_\_\_\_\_ like behavior.
- a) solid
  - b) liquid
  - c) combination of solid and liquid
  - d) neither solid nor liquid
16. Nylon threads are made of
- a) polyester polymer
  - b) polyamide polymer

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- c) polyethylene polymer
  - d) polyvinyl polymer
17. Which of the following is a branched polymer?
- a) low density polymer
  - b) polyester
  - c) high density polymer
  - d) nylon
18. On the basis of mode of formation polymers can be classified:
- a) as addition polymers only
  - b) as condensation polymers only
  - c) as copolymers
  - d) as addition and condensation polymers
19. The process of heat softening, moulding and cooling to rigidity can be repeated for which plastics?
- a) thermoplastics
  - b) thermosetting plastics
  - c) both (a) and (b)
  - d) neither (a) nor (b)
20. Which of the following monomers form biodegradable polymers?
- a) 3-hydroxybutanoic acid + 3-hydroxypentanoic acid
  - b) Glycine + amino caproic acid
  - c) ethylene glycol + phthalic acid
  - d) both a and b
21. In addition polymer, monomer used is
- a) unsaturated compounds
  - b) saturated compounds
  - c) bifunctional saturated compounds

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- d) trifunctional saturated compounds
22. Polymer formation from monomer starts by
- a) the condensation reaction between monomers
  - b) the coordinate reaction between monomers
  - c) conversion of monomer to monomer ions by protons
  - d) hydrolysis of monomers
23. Which of the following statements is not correct for fibres?
- a) Fibres possess high tensile strength and high modulus
  - b) Fibres impart crystalline nature
  - c) Characteristic features of fibres are due to strong intermolecular forces like hydrogen bonding
  - d) All are correct
24. Which of the following does not undergo additional polymerization?
- a) vinyl chloride
  - b) butadiene
  - c) styrene
  - d) all of the above undergoes addition polymerizations
25. Which one of the following is not a condensation polymer?
- a) Dacron
  - b) Neoprene
  - c) Melamine
  - d) Glyptal
26. Which of the following statements is false?
- a) The repeat unit in natural rubber is isoprene.
  - b) Both starch and cellulose are polymers of glucose.
  - c) Artificial silk is derived from cellulose.
  - d) Nylon-66 is an example of elastomer.

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27. Of the following which one is classified as polyester polymer?
- a) Nylon-66
  - b) Terylene
  - c) Backelite
  - d) Melamine
28. Which polymers occur naturally?
- a) Starch and Nylon
  - b) Starch and Cellulose
  - c) Proteins and Nylon
  - d) Proteins and PVC
29. Bakelite is obtained from phenol by reacting with \_\_\_\_
- a) HCHO
  - b)  $(\text{CH}_2\text{OH})_2$
  - c)  $\text{CH}_3\text{CHO}$
  - d)  $\text{CH}_3\text{COCH}_3$
30. Which one of the following statements is not true?
- a) Natural rubber has the trans-configuration at every double bond
  - b) Buna-S is a copolymer of butadiene and styrene
  - c) Natural rubber is a 1, 4-polymer of isoprene
  - d) In vulcanization, the formation of sulphur bridges between different chains make rubber harder and stronger.